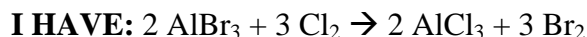


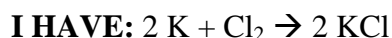
Chemical Reactions Loop Game



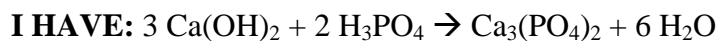
WHO HAS: Aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas?



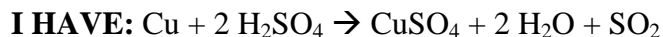
WHO HAS: Potassium metal and chlorine gas combine to form potassium chloride?



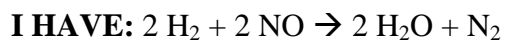
WHO HAS: Calcium hydroxide and phosphoric acid react to form calcium phosphate and water?



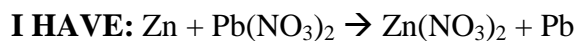
WHO HAS: Copper and sulfuric acid react to form copper (II) sulfate, water and sulfur dioxide?



WHO HAS: Hydrogen gas and nitrogen monoxide react to form water and nitrogen gas?



WHO HAS: Zinc and lead (II) nitrate react to form zinc (II) nitrate and lead?



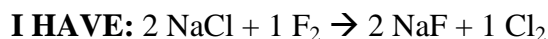
WHO HAS: Sodium phosphate and calcium chloride react to form calcium phosphate and sodium chloride?



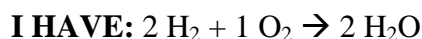
WHO HAS: Aluminum and hydrochloric acid react to form aluminum chloride and hydrogen gas?



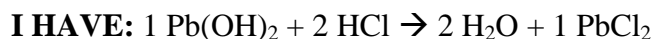
WHO HAS: Sodium chloride and fluorine gas react to form sodium fluoride and chlorine gas?



WHO HAS: Hydrogen and oxygen gases react to form water?



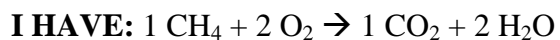
WHO HAS: Lead (II) hydroxide reacts with hydrochloric acid to form lead (II) chloride and water?



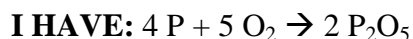
WHO HAS: Aluminum bromide and potassium sulfate react to form potassium bromide and aluminum sulfate?



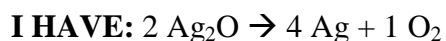
WHO HAS: Methane (CH_4) burns in oxygen to produce carbon dioxide and water?



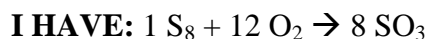
WHO HAS: Phosphorus and oxygen react to form diphosphorus pentaphosphide?



WHO HAS: Silver (I) oxide breaks down into silver and oxygen gas?



WHO HAS: Sulfur burns in the presence of oxygen to produce sulfur trioxide?



WHO HAS: Potassium reacts with magnesium bromide to produce potassium bromide and magnesium metal?

I HAVE: $1 \text{ K} + 1 \text{ MgBr} \rightarrow 1 \text{ KBr} + 1 \text{ Mg}$

WHO HAS: Sodium and water react to form sodium hydroxide and hydrogen gas?

I HAVE: $2 \text{ Na} + 2 \text{ H}_2\text{O} \rightarrow 2 \text{ NaOH} + 1 \text{ H}_2$

WHO HAS: Sulfuric acid and sodium nitrite react to form nitrous acid and sodium sulfate?

I HAVE: $1 \text{ H}_2\text{SO}_4 + 2 \text{ NaNO}_2 \rightarrow 2 \text{ HNO}_2 + 1 \text{ Na}_2\text{SO}_4$

WHO HAS: Mercury oxide breaks down into mercury and oxygen?

I HAVE: $2 \text{ HgO} \rightarrow 2 \text{ Hg} + 1 \text{ O}_2$

WHO HAS: Calcium burns in the presence of oxygen to produce calcium oxide?

I HAVE: $2 \text{ Ca} + 1 \text{ O}_2 \rightarrow 2 \text{ CaO}$

WHO HAS: Solid copper reacts with chlorine gas to produce copper (II) chloride?

I HAVE: $\text{Cu} + \text{Cl}_2 \rightarrow \text{CuCl}_2$

WHO HAS: Lead (II) sulfate reacts with iron (II) chloride to produce iron (II) sulfate and lead (II) chloride?